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Chemistry Name: ChemThink: Covalent Bonding

1. Covalent bonding forms when atoms are _____ electrons. 2. When two atoms get close enough, the nucleus attracts the other atom's (Protons / Neutrons / Electrons). 3. Before bonding, the atom's electrons spend most of their time _____ the nuclei of each atom

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In this Chemthink tutorial, you will explore covalent (molecular) bonding and take a short quiz. Topics include: why hydrogen atoms would attract when they share electrons. the relationship between distance, potential energy, and force between two atoms. naming covalent compounds. This new HTML5 version was built from the ground up to run on iPads, mobile phones, Chromebooks, and real computers in any HTML5-capable browser.

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Thank you for pointing that out! I just updated slides 24/27 and 25/27. I'm going to hold off on pushing out the update until the next version of ChemThink, which should be available within the next two weeks and include both "Covalent Bonding" and "Molecular Shapes" (VSEPR and Lewis Dot notation). I'm getting close! ?-Chris

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Quizlet

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