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Answers
1- every sample has exact same characteristic physical and chemical properties used to identify substance 2- every sample of a pure substance has exact same composition with the same makeup (i.e.: pure water is always 11.2%

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H and 88.8% oxygen
by mass

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This topic covers 1.2
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Structure and 1.3.1
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and boiling trends
only) Bonding,
Structure and

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Choice Identify the
choice that best
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LETTER of the correct
answer in the blank.

_____ 1. Inorganic
chemistry is the study
of a. non-carbon
related compounds.

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$2 = 253.80 \text{ g mol}^{-1}$
mass of $\text{I}_2 = 1.25 \times 10^{-3} \times 253.80$, i.e.
 0.317 g 16 a molar
mass of $\text{PbI}_2 =$
 460.99 0.1270 moles
of $\text{PbI}_2 = 460.99 =$
 $2.755 \times 10^{-4} \text{ mol b}$
 $\text{Pb}(\text{NO}_3)_2(\text{aq}) + \text{MI}$
 $2(\text{aq}) \rightarrow \text{PbI}_2(\text{s}) +$
 $\text{M}(\text{NO}_3)_2(\text{aq})$

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