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## Chapter 9 part 10 (FINALE)

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Answers

CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW  
SHORT ANSWER Answer the following questions in the  
space provided. 1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  a. What is the value of the  
coefficient x in this equation? 40.07 g/mol b. What is the  
molar mass of  $C_3H_4$ ? 2 mol  $O_2$ :1 mol  $H_2O$  c. What is the  
mole ratio of  $O_2$  to  $H$

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Modern Chemistry 77 Stoichiometry CHAPTER 9 REVIEW  
Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

1. \_\_\_\_\_ The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0

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mol of  $N_2$  are mixed with 12.0 mol of H

## ~~CHAPTER 9 REVIEW~~ Stoichiometry

Stoichiometry b. Theoretically, how many moles of  $NH_3$  will be produced? PROBLEMS Write the answer on the line to the left, Show all your work in the space provided. 1 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of  $N_2$  are mixed with 12.0 mol of  $H_2$  according to the ...

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Stoichiometry. SECTION 2. PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.  
1. The following equation represents a laboratory preparation for oxygen gas: ... CHAPTER 9 REVIEW ...

## ~~CHAPTER 9 REVIEW~~

Chapter 9: Standard Review Worksheet 1. Answers will vary. An example is included below:  $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2$



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(g) This describes the decomposition reaction of hydrogen peroxide. Microscopic: Two molecules of hydrogen peroxide (in aqueous solution) decompose to produce two molecules of liquid water and one molecule of oxygen gas.

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elements in compounds Reaction Stoichiometry - Involves mass relationships between reactants and products in a chemical reaction. I. Reaction Stoichiometry Problems A. Four problem Types, One Common Solution.

## ~~Chapter 9 Stoichiometry~~

Chapter 9 Review Stoichiometry Answers CHAPTER 9  
REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided. 1.

Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$   
4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar

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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW

SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  4 a.

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